



## **MATERIAL SAFETY** DATA SHEET



I - PRODUCT IDENTIFICATION AND USE MSDS ID: 0056						
PRODUCT NAME: 19% SODIUM HYPOCHLORIT	E					
USE: Liquid Bleach						
SUPPLIER: JohnsonDiversey Canada, Inc. 2401 Bristol Circle Oakville Ontario, L6H 6P1, Canada		EMERGENCY PHONE: 1-800-668-7171				
WHMIS CLASSIFICATION: D2B E CHEMICAL FAMILY: Chlorinated alkali		TRADE NAME / SYNONYMS:       not applicable         CHEMICAL NAME:       Sodium hypochlorite solution				
II - HA	ZARDOUS	INGF	REDIENTS			
HAZARDOUS INGREDIENT	% w/w		CAS #	LD50 / LC50	Route / Species	
ium hypochlorite 10-3			007681-52-9	not available	n/a	
III - HANDLING	AND DISP	OSA	PROCEDURES			
Respiratory:       If mists are generated, use NIOSH approver mask         SPECIAL HANDLING PROCEDURES AND EQUIPMENT:         VENTILATION REQUIREMENTS:       mechanical exhaust         INCOMPATIBILITY (Material to Avoid):       Acids, oxidizable models         SPILL PROCEDURES:       Contain the spill. Do not allow the or dilute hydrogen peroxide. Mop models         WASTE DISPOSAL:       Dispose according to municipal, provin	Avoid eye an naterials, amr spilled produc up or soak up ncial and feder	imperr d skin o monia, ct to go with al ral regu	urea, other nitrogenou to drain. Neutralize v osorbent clay for dispo ilations.	s materials, and meta vith sodium sulfite, sod sal. Wash spill area v	dium bisulphite, with large	
STORAGE / SHIPPING REQUIREMENT: Store in a cool dry				om oxidizable materia	ls. UN1791	
	HYSICAL P	ROP	ERTIES			
APPEARANCE / ODOUR:       Clear, yellow-green liquid - chlorine odour         S.G. / BULK DENSITY(g/cc):       1.25 ± 0.01         VAPOUR PRESSURE (mmHg):       17.5         ODOUR THRESHOLD:       not available         FREEZING POINT:       -25°C		pH:       (concentrate): 12.0         VAPOUR DENSITY (air=1):       not applicable         BOILING POINT:       approx. 100°C         PERCENT VOLATILE:       85%				
SOLUBILITY IN WATER: soluble V - TOXICOLOGICA			EVAPORATION RATE (water=1): not available			
EFFECTS OF ACUTE EXPOSURE TO MATERIAL:         EYES:       Corrosive. May cause severe irritation. May cause severe irritation. May cause severe irritation. May cause severe irritation. May cause independent of the cause severe irritation of the cause convulsions, coma and death.         INHALATION:       May cause irritation of the nose and throat, conversional cause irritation is the nose interversional cause irritation is the nose irritatirritation is the nose irritation irritation irritation irritation	ause permane ause tempora digestive tract.	ent dam ry or pe . May c	nage if not treated pror ermanent damage if no ause permanent dama	ot treated promptly.	nptly. May lead	

LD50 (calculated): not available	LC50 (calculated): not available					
OTHER TOXIC EFFECTS: none known						
EFFECTS OF CHRONIC none known EXPOSURE TO MATERIAL:						
VI - FIRST AID MEASURES						
<b>EYES:</b> Flush eyes with plenty of water for at least 15 minutes. Hold eye lids open while rinsing. Contact a physician immediately.						
<b>SKIN:</b> Flush affected area thoroughly with water. If irritation persists, contact a physician.						
<b>INGESTION:</b> Drink large volumes of water . Never give anything by mouth to an unconscious person. Do not induce vomiting. Contact a physician immediately.						
INHALATION: Remove patient to fresh air. If breathing difficulty occurs, get medical attention.						
FLAMMABLE: No						
FLASH POINT, °C: not applicable	AUTO IGNITION TEMPERATURE, °C: N/Ap					
	n Dioxide [X] Foam [X] Other [ ]					
SPECIAL FIRE FIGHTING PROCEDURES: Wear self-contained breathing apparatus and full protective equipment.						
HAZARDOUS COMBUSTION PRODUCTS: chlorine gas						
EXPLOSIVE SENSITIVITY TO: not applicable						
VIII - REACTIVITY DATA						
STABILITY: Stable [] Unstable [X]						
<b>CONDITIONS TO AVOID:</b> Decomposes slowly. Do not expose to temperatures above 40°C (104°F), sunlight, or metals.						
INCOMPATIBILITY (Material to Avoid) : Acids, oxidizable materials, ammonia, urea, metals.						
HAZARDOUS DECOMPOSITION PRODUCTS: Chlorine gas is released by contact with acids; oxygen is released by contact with metals. Contact with ammonia and urea produce nitrogen gas and chloramines. Contact with oxidizable materials produces heat which may generate chlorine gas.						
REACTIVITY: not applicable						
IX - MSDS PR	EPARATION					
SOURCES USED: RTECS	PREPARED BY: JohnsonDiversey Canada, Inc.					
	Regulatory Department					
PREPARATION DATE: April 15, 2003	Industrial Division Phone (905) 829-1200					
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